## LITHIUM N-BENZYLTRIMETHYLSILYLAMIDE (LSA): A NEW REAGENT FOR CONJUGATE ADDITION - ENOLATE TRAPPING REACTIONS

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Abstract-Lithium N-benzyltrimethylsilylamide (LSA) adds to crotonates in a 1,4-manner, though the reaction of ordinary lithium amides with  $\alpha,\beta$ -unsaturated carbonyl compounds is accompanied with a 1,2-addition and hydrogen abstraction at the y-position. The conjugate addition via LSA followed by enolate trapping with electrophiles produces the corresponding  $\alpha$ -substituted  $\beta$ -amino esters, which are, in turn, converted into  $\beta$ -lactams and  $\alpha$ -substituted  $\alpha,\beta$ -unsaturated esters.

Lithium amides  $R_2NLi$  are ordinarily used as strong bases for deprotonation of organic compounds. However, nucleophilic reactions of  $R_2NLi$ , such as conjugate addition to  $\alpha,\beta$ -unsaturated esters, have received little attention from a synthetic point of view. Here that lithium N-benzyl-trimethyl-silylamide [LiN(CH2Ph)SiMe3] (LSA)4 reacts with  $\alpha,\beta$ -unsaturated esters in a 1.4-manner to produce the corresponding  $\beta$ -amino ester enclates, which are trapped by electrophiles such as alkyl halides and aldehydes. The resulting  $\alpha$ -substituted  $\beta$ -amino esters can be converted into the  $\beta$ -lactams or into the  $\alpha$ -substituted  $\alpha,\beta$ -unsaturated esters (Scheme 1).

## RESULTS AND DISCUSSION

The reactions of methyl crotonate (1) with various Conjugate addition. kinds of lithium amides were investigated to elucidate the regioselectivity toward a, \$-unsaturated ester. The results are summarized in Table 1. LSA was an efficient reagents for the conjugate addition; by-products such as 3 and 4 were not detected. It was reported that LDA gave 2 in good yields without contamination of 3 and 4.  $^{1-3}$ Our experiments revealed that LDA produced significant amounts of 4 as a by-product; the n-octylated derivative of 4 was isolated since isolation of 4 itself was difficult owing to its volatile characteristics. Further, the isopropyl group of LDA can not be removed in the subsequent process and thus this reagent is not suitable for the preparation of β-lactams. The silylsubstituted amines such as Bn(t-BuMe,Si)NH,  ${\rm Bn(Ph_{9}MeSi)NH,\ Bn(Ph_{3}Si)NH,\ and\ Bn(t-BuPh_{2}Si)NH\ were\_prepared\ from\ the\ reaction}$ of benzylamine with the corresponding chlorosilanes.<sup>5</sup> The conjugate adduct 2 was obtained along with 3 via Bn(t-BuMe,Si)NLi or Bn(Ph,MeSi)NLi. Bn(Ph<sub>3</sub>Si)NLi or Bn(t-BuPh<sub>2</sub>Si)NLi was used, several unidentified materials were Other lithium amides such as BnNHLi and Bn2NLi, in which Bn group can be replaced by H in the subsequent process, also afforded both 1,4- and 1,2-The deprotonation reaction leading to 4 took place with (TMS) NLi.

Table 1. Reaction of methyl crotonate with lithium amides (Bn = CH<sub>2</sub>Ph)

| R <sub>2</sub> NL1            | isolated product, % |      |                  |  |
|-------------------------------|---------------------|------|------------------|--|
| <i>66</i>                     | 2                   | 3    | 4                |  |
| LDA(IPr <sub>2</sub> NLi)     | 44 <sup>b</sup>     | A.S. | 1 4 <sup>8</sup> | and the second s |
| LSA(Bn(TMS)NLI)               | 88                  | •    | Top.             |  |
| BnNHL i                       | 20                  | 60   | -                |  |
| Bn <sub>2</sub> NL i          | 18 <sup>b</sup>     | •    | ~                |  |
| Bn(t-BuMe <sub>2</sub> Si)NLi | 61                  | 7    | -                |  |
| Bn(Ph <sub>2</sub> MeSi)NLi   | 30                  | 6    |                  |  |
| (TMS) 2NL i                   |                     | ~    | 9 <sup>8</sup>   |  |

a) Instead of  $H_3O^+$ , octyl iodide was added. The octylated derivative of 4 was isolated in the indicated yields. b) Isolated as  $IPr_2N$  or  $Bn_2N$  adducts, respectively.

Conjugate addition-enolate trapping reaction. By using LSA, the conjugate addition to 1 followed by treatment with various electrophiles was studied. With n-octyl iodide, a 59:41 mixture of the anti (5a) and syn (5b) isomers was isolated in 81% yield. The isomer ratio depended upon the additives; the presence of IMPA gave a 89:11 mixture in 96% yield and the presence of TMEDA produced a 77:23 mixture in 78% yield. It should be noted that the TMS group of LSA is removed during work-up procedures. With methyl iodide, a 52:48 mixture of 6a and 6b was obtained in an essentially quantitative yield. With benzaldehyde, the adduct (7) was obtained in 97% yield; the ratio of four isomers (7a, b, c, d) was 57:24:17:2. Unfortunately, the stereochemistry of these isomers has not yet been determined.

B-Lactam synthesis. The anti 8-amino ester (5a) was converted into 8a in 90% yield via hydrolysis followed by dehydration [KOH/aqueous MeOH; PPh<sub>3</sub>,  $(PyS)_2/MeCN]$ . Quite similarly, 5b gave 8b in 74% yield. The stereochemistry of 8a and b, in turn the stereostructure of 5a and b, was determined by their H n.m.r. coupling constant: (8a)  $J_{3/4}$  2.0Hz; (8b)  $J_{3/4}$  5.3Hz.

 $\frac{\alpha-Alkylated}{7^{-9}} \text{ via quarternization-base treatment to produce the corresponding } \alpha-alkylated } \frac{3}{\alpha}-ansaturated esters.}$  N-Methylation of Sa with Mel/K<sub>2</sub>CO<sub>3</sub> followed by treatment with silica gel produced a 86 : 14 mixture of the E (9a) and Z (9b) isomers in 90% yield. Similar treatment of Sb again gave a 91 : 9 mixture of 9a and 9b in 86% yield. The three-step sequence, conjugate addition of LSA to 1 - enolate trapping with octyl iodide - \$\text{6}-elimination},

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afforded a 88: 12 mixture of 9a and 9b in 72% total yield. The conjugate addition of LDA to 1 followed by enolate trapping with octyl iodide and subsequent treatment with silica gel  $^{10}$  gave a 93: 7 mixture of 9a and 9b in 56% total yield. Therefore, the synthetic equivalent of the  $\alpha$ -carbanion of an  $\alpha,\beta$ -unsaturated ester can be generated by conjugate addition of a nitrogen nucleophile ( $R_2$ NLi); LDA, as well as LSA, is also useful for this purpose. Quite similarly, a 88: 12 mixture of 11a and 11b was obtained in 52% yield via the three-step sequence with LDA. The  $\alpha$ -benzylated derivative 10 (10a/10b = 87/13) was produced in 74% yield via LSA. Other  $\alpha,\beta$ -unsaturated esters such as ethyl 4-methyl-2-pentenoate  $^{11}$  and ethyl 2-decenoate also underwent the conjugate addition of LDA - trapping with MeI -  $\beta$ -elimination, giving 12 (E/Z = 95/5, 33%) and 13 (E/Z = 88/12, 69%), respectively. Consequently, the three-step sequence via LSA or LDA provides a new synthetic procedure for trisubstituted enoates.

Stereochemistry of intermediate enclate. The stereoselectivity of the conjugate addition - enclate trapping with octyl iodide depended upon the additives; the presence of HMPA increased the anti-isomer. To help clarify the origin of the anti-stereoselectivity, we intended to isolate the intermediate enclate. The conjugate adduct of LSA to 1 was trapped with Me<sub>3</sub>SiCl. It is well known that kinetic deprotonation of esters with lithium diskylamides produces the Z-enclate (14). Therefore, it was anticipated that E-15 would be produced predominantly in our case. The intermediate O-silylated ketene acetal 15 was isolated as a single isomer. We attempted to assign the E or Z configuration by using a NOE technique, but the attempt was unsuccessful. At present, we are unable to clarify the relation between the syn-anti stereoselectivity of 5-7 and the E-Z stereochemistry of 15.

## EXPERIMENTAL.

H NMR spectra were recorded with a Varian EM-390, XL-200, or Jeol GX-400 instruments with TMS as internal standard. IR spectra were recorded with a Hitachi 215 spectrophotometer. Mass spectra were recorded with a Hitachi M-52 or Jeol DX-303 spectrometer. Elemental analyses were performed by the Tohoku University Microanalytical Center.

Preparation of silylated amines. N-(Trimethylsilyl)benzylamine was

prepared according to the literature procedure; 14 1 H NMR (CCI, ) & 0.19 (s, 9), 0.45-1.08(br. 1), 4.00(d, J=7.9Hz, 2), 7.05-7.35(m, 5). Other amines were prepared by the procedure of Narula and Kapur. 5 N (t-Buty|dimethy|si|y|)benzy|amine. bp 92.5°C/2.5mmHg;  $^{1}$ H NMR (CCl $_{4}$ ) 5 0.17 (s, 6), 0.38-0.92 (m, 1), 1.05 (s, 9), 4.06 (d, J = 7.8Hz, 2), 7.12-7.38 (m. 5); IR (CCI<sub>A</sub>) 3040, 2970, 2940, 2870, 1650, 1500, 1470, 1405, 1260, 830cm<sup>-1</sup>; MS calcd for  $C_{13}H_{23}NS_1$  m/z 221.1600, found m/z 221.1609. N-(Diphenylmethylsilyl)benzylamine. bp 215°C/0.5mmHg; H NMR(CCl<sub>4</sub>) 6 0.56 (s, 3), 0.85-1.38 (br, 1), 3.95 (d, J\*7.5Hz, 2), 6.92-7.65 (m, 15); IR (CCI<sub>4</sub>) 3070, 3030, 1455, 1430, 1400, 1260, 1120, 700cm<sup>-1</sup>; MS calcd for  $C_{20}H_{21}NS1 \text{ m/z}$ 303.1443, found m/z 303.1482. N-(t Butyldiphenylsilyi)benzylamine. bp 233°C/0.5 mmHg; H NMR (CCl<sub>4</sub>) 6 0.93-1.30 (br, 1), 1.04 (s, 9), 3.90 (d, J+7.5Hz, 2), 6.93-7.73(m, 15); 1R (CC)<sub>4</sub>) 3080, 2940, 2865, 1470, 1455, 1430, 1405, 1390, 1110, 700cm<sup>-1</sup>; MS calcd for C<sub>23</sub>H<sub>27</sub>NS<sub>1</sub> m/z 345.1913, found m/z 345.1866. N (Triphenylsilyl)benzylamine. mp 85.0°C; H NMR (CCl<sub>4</sub>) 6 1.43 (bt, J=3.8Hz, 1), 4.05(d, J=3.8Hz, 2), 6.96-7.67(m, 20H); IR(KBr) 3390, 3075, 1430, 1405, 1120, 845, 740, 700cm<sup>-1</sup>. Anal. Calcd for C<sub>25</sub>H<sub>23</sub>NS1:C, 82.14; H, 6.34; N, 3.83. Found: C, 81.97; H, 6.46; N, 3.60.

Reaction of 1 with lithium amides. The reaction of LSA is representative. In a 20ml flask under Ar atmosphere were placed N-(TMS)-benzylamine (0.43lml, 2.2mmol) and dry THF (5ml). The flask was cooled to -78°C, and then Bulli-hexane solution (1.38ml, 2.2mmol) was added. The resulting solution was stirred for 30min, and then a THF (3ml) solution of 1 (0.212ml, 2mmol) was slowly added. The color changed to pale yellow. The stirring was continued for another 30min at this temperature. The reaction was quenched with a THF solution of HOAc. A saturated solution of NaHCO3 was added to make the solution basic. Extraction with ether, washing with brine, drying with  $K_2CO_3$ , condensation, and a chromatography by using silica gel gave the product.

Methyl 3-(benzylamino)butanoate (2). HNMR (CCl4) 5 i.ii (d, J=6.2Hz, 3), 1.50 (s, 1), 2.28(dd, J=6.2, 15.2Hz, 1), 2.36 (dd, J=6.2, 15.2Hz, 1), 3.05 (ddq,

 $\frac{\text{Methyl 3-(benzylamino)butanoste (2)}}{\text{(s, 1), 2.28(dd, J=6.2, 15.2Hz, 1), 2.36 (dd, J=6.2, 15.2Hz, 1), 3.05 (ddq, J=6.2, 6.3, 6.3Hz, 1), 3.59 (s, 3), 3.72 (s, 2), 7.00-7.28(m, 5); IR (nest) 3330, 1735, 1605, 1440, 1195, 1175, 740 cm<sup>-1</sup>; MS calcd for <math>C_{12}H_{17}NO_2$  m/z 207.1259, found m/z 207.1257.

N-Benzyl-2-butenamide (3). H NMR (CCl<sub>4</sub>)  $\delta$  1.83 (dd, J=1.7, 6.8 Hz, 3), 4.47 (d, J=5.9 Hz, 2), 5.43-5.83 (br, 1), 5.75 (dq, J=1.7, 16.5 Hz, 1), 6.83 (dq, J=6.8, 16.5 Hz, 1), 7.24(S, 5); IR (KBr) 3270, 1670, 1625, 1560, 1430, 970cm<sup>-1</sup>, MS calcd for C<sub>11</sub>H<sub>13</sub>NO m/z 175.0997, found m/z 175.0998.

 $\frac{\text{Methyl 3-(N, N-dibenzylamino)butanoste.}}{\text{3), 2.14 (dd, J=7.2, 13.8 Hz, 1), 2.53 (dd, J=7.2, 13.8 Hz, 1), 3.25 (ddq, J=6.2, 13.8, 13.8Hz, 1), 3.43 (s, 3), 3.38, 3.58(2d, J=13.5Hz, 4), 6.90-7.33 (m, 10); <math>IR(CCl_4)$  3045, 1745, 1458, 1195,  $700cm^{-1}$ ; MS calcd for  $C_{19}H_{23}NO_2$  m/z 297.1729, found m/z 297.1744.

for  $C_{11}H_{23}NO_2$  m/z 201.1729, found m/z 201.1727.

Methyl 2-octyl-3-butenoste.

1H NNR (CCl<sub>4</sub>) 5 0.75-1.02 (m, 3), 1.02-1.54 (bs, 14), 2.89 (m, 1), 3.61 (s, 3), 5.05 (m, 2), 5.68 (m, 1); IR (CCl<sub>4</sub>) 1735 (1455, 1435, 990, 920, 700cm<sup>-1</sup>; MS calcd for  $C_{13}H_{24}O_2$  m/z 212.1776, found m/z 2.37 (dd, J=7.6, 13.5Hz, 1), 3.07 (hep, J=6.6Hz, 2), 7.6 Hz, 1), 3.55 (s, 3); IR (CC1<sub>4</sub>) 1745, 1465, 1440, Methyl 3-(N, N-diisopropylamino)butanoste.

H NMR (CCI<sub>d</sub>) 6 0.96 (d, J=6.6Hz, 6), 1.03 (d, J=6.6Hz, 6), 2.14 (dd, J=7.0, 13.5Hz, 1), j=6.6Hz, 2), 3.39(ddq, j=6.5, 7.0, 1465, 1440, 1400, 1220cm<sup>-1</sup>; MS calcd

Methy | 3- (benzylamino) - 2-methylbutanoate. Mel (1.59ml, 25mmol) was added. isolated via the usual work-up procedure. of LSA was performed as described above. room temperature, and then quenched with aqueous  $\mathrm{NH_4Cl}$ . Conjugate addition - enolate trapping reactions. The reaction mixture was allowed to warm After stirring at -78°C for thr, H NMR (CDC) 3) (85 8 52 : 48 The conjugate additi The product was

7.00-7.28 (bs, 5); IR (neat) 3340, 3040, 2940, 2870, 1735, 1460, 1440, 1380 1200, 1165cm<sup>-1</sup>; MS calcd for  $C_{20}H_{33}NO_{2}$  m/z 319.2512, found m/z 319.2484. 5b (syn); <sup>1</sup>H NNR (CCl<sub>4</sub>) 6 0.74-0.97 (m, 3), 1.05 (d, J=6.0Hz, 3), 1.11-1.80 (m, 14), 1.11-1.80 (br, 1), 2.23-2.52 (m, 1), 2.76 (dq, J=5.3, 6.0Hz, 1), 3.60 (s, 3), 3.73 (s, 2), 7.00-7.30 (bs, 5); IR (CCl<sub>4</sub>) 3340, 3040, 2940, 28 1735, 1605, 1455, 1440, 1380, 1200, 1165cm<sup>-1</sup>; NS calcd for  $C_{20}H_{33}NO_{2}$  m/z 319.2512, found m/z 319.2490. (m, 1), 2.58(dq, ]=6.8, 6.9Hz, 1), 2.63 (dq, ]=5.0, 7.0 Hz, 1), 2.92 (dq, ]\*5.0, 6.4Hz, 1), 2.99 (dq, ]\*6.7, 6.8Hz, 1), 3.67 (s, 3), 3.68 (s, 3), 3.78 (2d, ]=13.2Hz, 2), 7.20-7.34 (m, 10); IR (CCl<sub>4</sub>) 3400, 1740, 1457, 1200cm<sup>-1</sup>; NS calcd for  $C_{13}H_{19}NO_2$  m/z 221.1416, found m/z 221.1413. Methyl 3-(benzylamino)-2-octylbutanoate. 5a (anti);  $^1H$  NNR (CCl<sub>4</sub>) 6 0.75-1.00 (m, 3), 1.05(d, ]=6.3Hz, 3), 1.12-1.72 (m, 14), 1.03-1.72 (br, 1), 2.13-2.41 (m, 1), 2.81 (dq, ]=6.3, 6.3Hz, 1), 3.60 (s, 3), 3.64, 3.80 (2d, ]=13.5Hz, 2), 7.00-7.28 (bs, 5); IR (neat) 3340, 3040, 2940, 2870, 1735, 1460, 1440, 1380, 1200 (1165cm<sup>-1</sup>). No calcd for C  $^1H$  NO  $^1H$  120 (200 1165cm<sup>-1</sup>). No calcd for C  $^1H$  NO  $^1H$  210 (250 1735, 1460, 1440, 1380, 1380) mixture of diastereomers 6a and b) 60.83-0.93 (m, 1), 1.07(d, j=6.7Hz,J=6.4Hz, ಀ 1.13 (d, j=7.0Hz, 3), 1.17 (d, j=6.9Hz, 3), 1.21-1.34 2940, 2870, 3

3), 3.74 (s, 2), 5.03 (d, ]=4.7Hz, 1), 7.17 (bs, 10); IR ( $CCI_4$ ) 3310, 3075, 3040, 2960, 1725, 1603, 1495, 1455, 1440cm<sup>-1</sup>; MS calcd for  $CI_9H_2sNO_3$  m/z 313.1678, found m/z 313.1687. 7d;  $^1H$  NMR ( $CCI_4$ ) 6 1.19 (d, ]=6.3Hz, 3), 2.60 (dd, ]=2.7, 4.2Hz, 1), 2.78 (dq, ]=2.7, 6.3Hz, 1), 3.53, 3.79 (2d, ]=12.2Hz, 2), 3.68 (s, 3), 5.16 (d, ]=4.2Hz, 1), 6.93-7.33 (m, 10); IR ( $CCI_4$ ) 3150, 3030, 2930, 1730, 1495, 1450, 1435, 1165, 695cm<sup>-1</sup>; MS calcd for C<sub>19</sub>H<sub>23</sub>NO<sub>3</sub> m/z 313.1678, found m/z 313.1690. 1.11 (d, J=6.2Hz, 3), 2.37 (dd, J=9.0, 9.0Hz, 1), 3.22 (S, 3), 3.24 (dq, J=6.2, 9.0 Hz, 1), 3.76, 3.88 (2d, J=12.6Hz, 2), 4.80 (d, J=9.0Hz, 1), 6.97-7.40 (m, 10); IR (CCl<sub>4</sub>) 3100, 3040, 2960, 1730, 1455, 1435, 700cm<sup>-1</sup>; MS celcd for C<sub>19</sub>H<sub>23</sub>NO<sub>3</sub> m/z 313.1678, found m/z 313.1699. 7c; <sup>1</sup>H NMR (CCl<sub>4</sub>) 6 1.14 (d, (7a, b, c) were isolated as a partition, (7a, b, c) were isolated as a partition, (7a, l), c) were isolated as a partition, (2.99 (dq, 7a; l) NMR (CCl<sub>4</sub>) & 1.20 (d, J=6.3Hz, 3), 2.60 (dd, J=3.5, 3.5Hz, 1), 2.99 (dq, J=3.5, 6.3Hz, 1), 3.02 (s, 3), 3.64, 3.95 (2d, J=13.4Hz, 2), 4.95 (d, J=3.5Hz, J=3.5, 6.3Hz, 1), 7.00-7.38 (m, 10); IR(CCl<sub>4</sub>) 3150, 2940, 1730, 1455, 1195, 1170cm<sup>-1</sup>; NS calcd for C.-H-NO, m/z 313.1678, found m/z 313.1681. 7b; <sup>1</sup>H NMR (CCl<sub>4</sub>) & Methyl 2-(1-benzylaminoethyl)-3-hydroxy-3-phenylpropanoate. b, c) were isolated as a pure form, though 7d was contaminated with 7a. 6.3Hz, 1), 3.02 (dq, J\*6.3, 6.5Hz, 1), 3.46 (s, Three isomers

MeOH (2.5ml),  $m H_{2O}$  (0.5ml) and KOH (2 pellets). Hydrolysis of 5 (128mg, 0.4mmol) was performed with The mixture was kept

40°C for 36hr. Neutralization with 1.4N HCl, followed by evaporation of MeOH and  $\rm H_2O$ , gave a white precipitate. The mixture of this acid,  $\rm CH_3CN$  (40ml),  $\rm Ph_3P$  (133.6mg, 0.48mmol), and  $\rm (PyS)_2$  (108.4mg, 0.48mmol) was refluxed for 2 days under nitrogen. Condensation followed by purification via a silica gel column chromatography by using hexane-ether- $\rm CH_2Cl_2$  (1:1:1) as an eluant produced 8.

 $\frac{\text{trans 1-Benzyl 4-methyl-3-octyl-2-azetidinone}}{\text{0.71-1.00 (m, 3), 1.18 (d, J=6.0Hz, 3), 1.07-1.78 (m, 14), 2.44-2.68 (m, 1), 3.09 (dq, J=2.0, 6.0Hz, 1), 3.93, 4.50 (2d, J=15.0Hz, 2), 7.19 (bs, 5); IR (neat) 2970, 2930, 2860, 1750, 1400, 730, 700cm <math>^{-1}$ ; MS calcd for  $C_{19}H_{29}NO \text{ m/z}$  287.2249, found m/z 287.2240. The cis-isomer (8b);  $^{-1}H$  NMR (CC1<sub>4</sub>) 6 0.74-1.00 (m, 3), 1.07 (d, J=6.3Hz, 3), 1.12-1.67 (m, 14H), 2.99 (dt, J=5.1, 5.3Hz, 1), 3.53 (dq, J=5.3, 6.3Hz, 1), 3.97, 4.48 (2d, J=15.0Hz, 2), 7.19 (bs, 5); IR (CC1<sub>4</sub>) 2920, 2850, 1735, 1445, 1365, 700cm  $^{-1}$ ; MS calcd for  $C_{19}H_{29}NO \text{ m/z}$  287.2249, found m/z 287.2252.

a-Alkylated a 18 - unsaturated ester. Preparation of 9a, b is representative. The a-octylated enoate 5 (158mg, 0.49mmol) was placed in a 10ml flask, and then MeOH (1.5ml) and  $K_2CO_3$  (0.35g) were added. The mixture was cooled to 0°C and then Mel (0.25ml, 4mmol) was added. The mixture was kept at room temperature for 15hr. Water was added and  $K_2^{\infty}$  was dissolved. Ether extraction gave the methylated derivative. This N-benzyl-N-methylamino derivative (133.4mg, 0.4mmol) was refluxed with  ${\rm SiO_2}$  (0.04-0.063mm, 270mg) in toluene (2.9ml). After 60hr, the B-elimination was completed. Methyl 2-octyl-2-butenoate (9). The trans-isomer (9a) was contaminated with the cis-isomer (9b). The spectral data was based on a 93:7 mixture.  $^{1}H$  NMR (CCI $_{4}$ )  $_{5}$  0.73-1.07 (m, 3), 1.23 (bs, 12), 1.79 (d, J=7.2Hz, 3), 2.10-2.37 (m, 2), 3.64 (s, 3), 6.69 (q, J-7.2Hz, 1); IR (CCI<sub>4</sub>) 2960, 2935, 2860, 1715, 1460, 1440cm  $^{1}$ ; MS calcd for  $C_{13}H_{24}O_{2}$  m/z 212.1776, found m/z 212.1759. The trans-configuration was assigned by the chemical shifts of the olefinic Me and  $\mathrm{H}^{15}$  (1.79 and 6.69, respectively); the corresponding chemical shifts of the cis-isomer appeared at 1.92 (dt, j=1.2, 7.2Hz) and 5.87 (q, j=7.2Hz). Further, photochemical irradiation of a 93:7 mixture of 9a and b produced a 50 : 50 mixture of both isomers.

Methyl 2-ethylidene-5-methyl-4-hexenoate (11). HNMR (CCl<sub>4</sub>) 8 1.63 (bs, 6) 1.80 (d, J=7.1Hz, 3), 2.93 (bd, J=6.9Hz, 2), 3.64 (s, 3), 4.92 (m, 1), 6.68 (q, 1, =CH), 5.87 (m, the cis isomer); IR (neat) 2930, 2860, 1720, 1650, 1440, 1285, 1235, 850cm<sup>-1</sup>; MS calcd for  $C_{10}H_{16}O_{2}$  m/z 168.1150, found m/z 168.1154. Ethyl 2,4-dimethyl-2-pentenoate (12). The <sup>1</sup>H NMR spectra was in good agreement with the reported data.

O-Silylated ketene acetal (15). After the conjugate addition of LSA to 1, 1 eq of TMSCI was added at -78°C. The reaction was allowed to warm

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to room temperature. The solvent was removed under vacuum. The ketene acetal was obtained along with LiCl.  $^1H$  NMR (THF-d<sub>8</sub> at -78°C) 0.13 (s, 9, OSiMe<sub>3</sub>), 0.24 (s, 9, NSiMe<sub>3</sub>), 1.01 (d, J=6.8Hz, 3, OH<sub>3</sub>ON), 3.29 (s, 3, OMe), 3.58 (d, J=9.5Hz, 1, CHON), 3.93-4.10 (m, 1, CHN), 4.03 (s, 2, CH<sub>2</sub>Ph), 7.13-7.36 (m, 5, Ph).

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